

UNIVERSIDADE DE SÃO PAULO
FACULDADE DE ODONTOLOGIA DE BAURU

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In situ effect of a proanthocyanidin mouthrinse on dentin submitted to erosion

**Efeito *in situ* de uma solução de proantocianidina sobre a dentina
submetida à erosão dentária**

BAURU

2017

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Dissertação constituída por artigo apresentada a Faculdade de Odontologia de Bauru da Universidade de São Paulo para obtenção do título de Mestre em Ciências no Programa de Ciências Odontológicas Aplicadas, na área de concentração Dentística.

Orientador: Prof. Dr. Heitor Marques Honório

BAURU

2017

Cardoso, Fabrícia

C179e *In situ effect of a proanthocyanidin mouthrinse on dentin submitted to erosion/* Fabrícia Cardoso. – Bauru, 2017.

56 p. : il. ; 31cm.

Dissertação (Mestrado) – Faculdade de Odontologia de Bauru. Universidade de São Paulo

Orientador: Prof. Dr. Heitor Marques Honório

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Assinatura:

Comitê de Ética da FOB-USP

Protocolo nº: CAAE49813015.7.0000.5417

Data: 04/12/2015

FOLHA DE APROVAÇÃO

DEDICATÓRIA

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Dedico esta dissertação aos meus pais, Neusa e Osni, e à minha irmã Estela (Johley) não só por me incentivarem, mas por terem certeza que eu chegaria até aqui. Obrigada por acreditarem e compartilharem os mesmos sonhos que eu.

AGRADECIMENTOS

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À Deus, presente em cada passo, cada suspiro e em cada piscar de olhos. Por me permitir sentir sua grandiosa presença em minha vida, por ser minha fortaleza, por se fazer notar nos mínimos detalhes. Por me ensinar a combater o bom combate e me guiar pelos caminhos que jamais trilharia sozinha. Por atender os anseios do meu coração. Por guardar-me como a menina dos olhos e sempre me proteger na sombra das tuas asas. Por ser minha essência vital.

Aos meus pais, Neusa e Osni, por segurarem todas as minhas lágrimas durante esta caminhada, me incentivando, me motivando e me cobrindo de amor. Por vibrarem de felicidade com meus progressos e a cada pequena conquista. Obrigada por acreditarem com todas as forças que eu superaria os vários obstáculos, inclusive nos momentos em que nem eu mesma acreditava. Deus não poderia ter me concedido maior benção do que ser filha de vocês. Lutamos juntos por este sonho! Essa vitória é nossa! Vocês são minha essência vital!!!

À minha querida Johley, meu alicerce, meu exemplo de disciplina e dedicação. Obrigada por me dar forças e me encorajar a seguir adiante. Por me fazer acreditar que sou melhor do que penso ser. Por confiar em cada escolha, cada passo. Obrigada pelo brilho nos olhos cada vez que fala das minhas conquistas. Deus não poderia ter me concedido maior benção do que ser sua irmã. Lutamos juntas por este sonho! Essa vitória é nossa! Você é minha essência vital!!!

Aos meus familiares, por compreenderem minhas ausências durante esses dois anos e torcerem por mim incondicionalmente.

Ao meu orientador, professor **Heitor**, que me recebeu de braços abertos e com uma humildade invejável desde o meu primeiro dia na FOB. Obrigada por pacientemente entender minhas limitações e me ajudar a superá-las. Por confiar em mim uma pesquisa de tal complexidade. Por transmitir conhecimentos com simplicidade, por incentivar novas ideias, por aconselhar como um amigo. Sua paixão pela profissão me contagia e me fez perceber que sim, escolhi o caminho certo. É uma honra dizer que sou sua orientada de mestrado! Que Deus possibilite a realização de todos os seus sonhos! Obrigada por tudo!

À Faculdade de Odontologia de Bauru – USP, por permitir a realização de um sonho.

À Angélica Feltrin pela amizade desde o início do mestrado. Por se sensibilizar com minhas dificuldades e sempre estar disposta a ajudar. Por me mostrar diferentes visões sobre a vida, pelos cuidados de mãe nos domingos, pelas sessões cinema e pela companhia toda vez que eu precisava correr atrás de comida devido à hipoglicemia. Minha dupla de mestrado e amiga pra toda vida. Obrigada por tudo!

À Ana Paula Boteon (sempre Sis) pela contribuição indispensável na parte experimental da minha pesquisa. Pelos conselhos, pelo ombro amigo e pela companhia no bandex. Pelas comidas requintadas que aprendeu na internet. Por ser essa pessoa de coração grandioso e por me presentear com sua amizade (e que ela seja eterna)! Obrigada por tudo!

Às voluntárias da minha pesquisa, por imensa paciência e dedicação com cada ciclagem. Por entenderem a magnitude do meu trabalho e atenderem meu pedido com seriedade. Sem vocês minha pesquisa não existiria! Muito obrigada!

À **Mayara Narimatsu** e **Lorena Garrido**, que me incentivaram e ajudaram em todo o processo de estudo para a prova do mestrado. Ajuda genuína e despretenciosa que admiro e levo como exemplo. Se estou aqui hoje, devo muito à vocês duas!

À **Aline Debastiani** e **Danielle Uehara** pela companhia em Bauru nestes dois anos. Vocês tornaram meus finais de semana menos tristes e menos solitários. Obrigada pelas conversas sérias e pelas conversas nem tão sérias assim. Por aceitarem o desafio de serem minhas futuras madrinhas! Vocês são preciosas pra mim!

À **Andressa Martins, Hellen Kurahashi, Tatiane Martino** e **Wilson Cirilo** por me darem forças durante o mestrado, por se alegrarem com as minhas vitórias, por estarem perto mesmo estando tão longe. Vocês são presentes que a graduação me deu e que Deus se encarregou de manter na minha vida! Obrigada pela linda amizade que construímos durante todos esses anos!

À todos os **alunos de graduação da FOB/USP**, especialmente à **turma LIV**. Obrigada por me receberem de braços abertos, com carinho e respeito. Por possibilitarem lindos laços de amizade. Por permitirem que eu tentasse passar tudo que sei e aprender mais ainda com cada um de vocês.

À todos os **professores do Departamento de Dentística**, especialmente ao professor **Rafael Mondelli** e à professora **Linda Wang**, pelas inúmeras chances de crescer como profissional e como ser humano. Pelas dicas, conselhos e críticas. Meu intuito sempre foi terminar o mestrado completamente diferente de quando entrei, absorvendo o máximo que conseguisse e aprendendo mais a cada dia.

Graças à vocês, isso foi possível! Muito obrigada!

Ao meu ex-professor de Dentística, **Anselmo Mariotto**, por despertar o meu amor por essa área linda da Odontologia! Pelos ensinamentos durante a graduação e por continuar acrescentando na minha vida profissional. Pela oportunidade de trabalharmos juntos, saiba que é uma honra!

Ao professor **José Mondelli**, que demorou pra decorar meu nome, mas que se tornou um grande amigo. Por fazer o melhor cafézinho da FOB, por sempre me receber em sua sala com um grandioso sorriso e por tentar (incansavelmente) ser meu cupido.

À **Maristela**, por todo esclarecimento e ajuda com o Comitê de Ética.

À **Tamires** (Catota), **Tamires** (Bueno), **Bianca** (Geri) e **Marian** (Piper) pela valiosa ajuda durante toda parte experimental da minha pesquisa. Além disso, pelos agradáveis momentos e muitas risadas até altas horas nos laboratórios. A alegria torna o trabalho muito mais leve! Obrigada!

Ao **Alcides** pela ajuda no laboratório durante a execução da minha pesquisa.

À todos os **colegas de pós-graduação**, pelos momentos de alegria e aprendizado compartilhados.

Aos funcionários do Departamento de Dentística e Materiais Odontológicos pelo acolhimento e ajuda com a parte burocrática desses dois anos de mestrado.

Aos funcionários do setor de limpeza, de alimentação e de segurança da FOB/USP por trabalharem com amor e dedicação, proporcionando um ambiente digno e de respeito.

Ao Conselho Nacional de Desenvolvimento Científico e Tecnológico pela concessão de bolsa de mestrado.

E tudo o que pedires em oração, crendo, o recebereis."

Mateus 22:21

ABSTRACT

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***In situ* effect of a proanthocyanidin mouthrinse on dentin submitted to erosion**

The organic matrix is of great importance in the process of loss of dental tissue because it functions as a barrier that prevents the diffusion of the acids to the tissues. For its degradation to be avoided, some agents have been tested in an attempt to inhibit the MMPs, the enzymes responsible for this process. Proanthocyanidin has been shown to be efficient in inhibiting these enzymes and therefore the aim of this study *in situ* was to evaluate the protective effect of a mouthrinse based on proanthocyanidin applied on the dentin submitted to erosion. Seven volunteers wore 2 palatal devices in each phase. The groups under study were: First Phase/ G1 - Erosive challenge with acid drink (Coca-cola®) before dentin treatment with 10% proanthocyanidin mouthrinse (neutralized to pH 7.0, Experimental group 1), G2 - Erosive challenge with acid drink (Coca-cola®) before dentin treatment with 10% proanthocyanidin mouthrinse (without neutralization, pH 3.0, Experimental group 2). Second Phase/ G3 - Erosive challenge with acid drink (Coca-cola®) before dentin treatment with 0.12% Chlorhexidine mouthrinse (pH 7.0, Positive control group), G4 - Erosive challenge with acid drink (Coca-cola®) with no previous treatment (Negative control group). Treatments with different mouthrinses were applied once after the second erosive challenge, for 5 minutes. Volunteers continuously wore the oral appliance and for the erosive cycling, each device was immersed into the 32 ml of acid beverage, 3 times a day for 5 minutes during 5 days. Profilometry was used to quantify the dentin loss (μm). Data were analyzed by Repeated measures ANOVA followed by LSD Fishers's test ($p<0.05$). G1 ($1.17^{\text{a}} \pm 0.69$) and G3 ($1.22^{\text{a}} \pm 0.25$) showed significant lower wear values with no statistical difference between them. There was also no significant differences between G2 ($2.99^{\text{b}} \pm 1.15$) and G4 ($2.29^{\text{b}} \pm 1.13$) resulting in more wear when compared to others groups. This study suggest that 10% neutralized proanthocyanidin mouthrinse could be a good strategy to diminish dentin wear progression.

Keywords: Proanthocyanidin; Tooth wear; Dentin; Erosion.

RESUMO

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Efeito *in situ* de uma solução de proantocianidina sobre a dentina submetida à erosão dentária

A matriz orgânica é de grande importância no processo de perda de tecido dental porque funciona como uma barreira que impede a difusão dos ácidos para os tecidos. Para evitar a sua degradação, alguns agentes foram testados numa tentativa de inibir as MMPs, enzimas responsáveis por este processo. A proantocianidina mostrou ser eficiente na inibição das mesmas e, portanto, o objetivo deste estudo *in situ* foi avaliar o efeito protetor de um enxaguatório com base em proantocianidina aplicada na dentina submetida à erosão. Este estudo cruzado simples-cego foi realizado em 2 fases de 5 dias cada. Sete voluntários usaram 2 dispositivos palatinos em cada fase. Os grupos estudados foram: Primeira Fase / G1 - Desafio Erosivo com bebida ácida (Coca-cola®) antes do tratamento da dentina com enxaguatório bucal de proantocianidina 10% (pH neutralizado a 7,0, Grupo Experimental 1), G2 - Desafio Erosivo com bebida ácida -cola®) antes do tratamento na dentina com enxaguatório bucal de proantocianidina a 10% (sem neutralização, pH 3,0, grupo experimental 2). Segunda Fase / G3 - Desafio erosivo com bebida ácida (Coca-cola®) antes do tratamento na dentina com 0,12% de enxaguatório bucal de Clorhexidina (pH 7,0, controle positivo), G4 - Desafio erosivo com bebida ácida (Coca-cola®) sem tratamento (controle negativo). Os tratamentos com diferentes enxaguatórios foram aplicados uma vez após o segundo desafio erosivo, durante 5 minutos. Os voluntários continuamente usavam o aparelho oral e para o ciclo erosivo, cada dispositivo foi imerso em 150 ml de bebida ácida, 3 vezes ao dia durante 5 minutos durante 5 dias. A Perfilometria foi utilizada para quantificar a perda de dentina (μm). Os dados foram analisados por ANOVA de medidas repetidas seguido por teste de Fishers LSD ($p < 0,05$). G1 ($1,17^{\text{a}} \pm 0,69$) e G3 ($1,22^{\text{a}} \pm 0,25$) mostraram valores de desgaste significativamente menores, sem diferença estatística entre eles. Também não houve diferenças significativas entre G2 ($2,99^{\text{b}} \pm 1,15$) e G4 ($2,29^{\text{b}} \pm 1,13$), resultando em maior desgaste quando comparado a outros grupos. Este estudo sugere que o enxaguatório bucal de proantocianidina 10% neutralizada poderia ser uma boa estratégia para diminuir a progressão do desgaste dentinário.

Palavras-chave: Proantocianidina. Desgaste dentário. Dentina. Erosão.

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1 INTRODUCTION

1 INTRODUCTION

Dental caries, which for some decades was the major concern of dentists, have shown a decrease in incidence in developed and underdeveloped countries. Therefore, non-carious lesions have become more frequent and among them, tooth wear (MAHONEY, KILPATRICK, 2003; HUYSMANS, CHEW, ELLWOOD, 2011; AL-ZAREA, 2012; MOAZZEZ, BARTLETT, 2014).

The tooth wear is subdivided into three categories: attrition, abrasion and erosion (BARTLETT, 2006; GANSS, 2006; HARA, LUSSI, ZERO, 2006; WEST, JOINER, 2014). Attrition is the superficial loss of dental structure due to dental contacts involving occlusal and incisal surfaces during swallowing, phonation and in cases of bruxism . Abrasion is caused by mechanical forces, but not related to the occlusion, among which we can highlight the technique of toothbrushing the type of brush used and the presence of abrasive in the toothpaste (ECCLES, 1982; IMFELD, 1996; MAGALHÃES et al, 2009a, RIOS et al, 2006; RIOS et al, 2007; WEST, JOINER, 2014).

Erosion presents a more complex pathophysiology, characterized as chronic and chemical erosion caused by acids without the involvement of microorganisms. It may occur due to intrinsic or extrinsic mechanisms (LUSSI, JAEGGI, ZERO, 2004; KAIDONIS, 2012; CARVALHO, 2014; LUSSI, GANSS, 2014).

Intrinsic erosion is resulted from the action of hydrochloric acid present in the stomach, which is found in the oral cavity through vomiting (anorexia and bulimia), gastric content regurgitation, recurrent reflux or gastric dysfunction (SCHEUTZEL, 1996; ZERO, 1996; LUSSI, JAEGGI, ZERO, 2004; AMAECHI, HIGHAM, 2005; BARBOSA et al, 2009; UHLEN et al, 2014). Extrinsic erosion occurs due to excessive consumption of beverages and foods with a high level of acidity or environmental exposure to acidic agents (ZERO, 1996; CHEHAL et al, 2009; ISAKSSON et al, 2014; JAEGGI, LUSSI, 2006).

Studies have shown that the evolution of the disease brings to the patient discomforts such as pain and hypersensitivity (O'SULLIVAN et al, 1998; RYTOMAA et al, 1998; RIOS et al, 2007). Because of that, it is of paramount importance that preventive methods be employed in order to minimize such symptoms, since the more advanced stage of

the disease demands more time and greater cost to the patient (SERRA, MESSIAS, TURSSI, 2009).

For the curative and preventive measures to be effective, it is necessary to know the different composition of the structures involved in the disease process, in this case enamel and dentin, as well as its behavior in the evolution of this type of lesion.

Approximately 85% of dental enamel is composed of mineral, in the form of highly substitutable hydroxyapatite crystals arranged in prisms, the remainder composed by proteins and water. In situations where the pH is mainly below 4.5 these crystals are easily solubilized by the acids, making the dental surface softened, which results in the loss of its structural component, characterizing tooth wear (BUZALAF et al, 2010, HONÓRIO et al, 2008; RIOS et al, 2008; HONÓRIO et al, 2010; LUSSI A, GANSS, 2014; SIERPINSKA et al, 2013).

The dentin is composed of inorganic, organic material and water. The inorganic matrix is represented by replaceable hydroxyapatite crystals, poor in calcium, phosphate, but rich in magnesium and carbonate when compared to enamel hydroxyapatite (SILVERSTONE, HICKS, 1985). These crystals, located between and on collagen fibrils, are also smaller in size when compared to those found in dental enamel, allowing them a larger surface area which makes them more vulnerable to acid attack (SILVERSTONE, HICKS, 1985).

The organic matrix is composed of 90% collagen fibrils (type I collagen) and the remaining 10% is composed of dentin, proteoglycan and glycosaminoglycan phosphoproteins (SILVERSTONE, HICKS, 1985; PASHLEY et al, 2004; MAGALHÃES et al, 2009a). Collagen fibrils decrease the rate of demineralization when present in a larger quantity, because the presence of collagen makes difficult the acid diffusion into the tissue, thus minimizing the erosion process (KLONT, TEN CATE, 1991; KLETER et al, 1994). Dentin also contains enzymes called matrix metalloproteinases (MMPs) that are activated by a drop in pH below 4.5. However, degradation of the dentin organic matrix is only possible after neutralization of saliva pH, since, although these enzymes are activated at acidic pH, they do not have the capacity to degrade the dentin matrix at the same pH (TJADERHANE et al, 1998; STROBEL, HELLWIG, 2015).

Natural enzymatic inhibitors have been increasingly studied by researchers (CASTELLAN et al, 2010; ISAKSSON et al, 2014) because they present lower toxicity when compared with synthetic products and minimal side effects (SABINO et al 1999; YE et al, 1999). Green tea has been shown to be effective in reducing dentin wear under erosive and/or

abrasive conditions (KATO et al, 2009). The green tea extract, epigallocatechin 3-gallate (EGCg), was added to a gel of topical application, which showed to be able to reduce the wear in dentin significantly, compared to fluoride (KATO et al, 2010). In addition, because it is an inhibitor of MMPs, this extract led to significantly lower concentrations of hydroxyproline compared to other treatments, proving its preventive effect on dentin erosion (KATO et al, 2012).

In research on dental caries and periodontal disease, some benefits related to polyphenols from Cranberry juice (*Vaccinium macrocarpon*) or extracted from the fruit itself have been scientifically proven. EGCg and Proanthocyanidins have become targets of studies, since they have the capacity to reduce the inflammatory response, as well as the production and activity of proteolytic enzymes (MMPs) that favor the destruction of the extracellular matrix in periodontal disease (LA, HOWELL, GRENIER, 2009; BONIFAIT, GRENIER, 2010). Several studies have demonstrated beneficial properties for the treatment and prevention of caries and periodontal disease (FEGHALI et al, 2012; GAZZANI, DAGLIA, PAPETTI, 2012), which would probably be related to the inhibition of the production and activity of matrix metalloproteinases (LA, HOWELL, GRENIER, 2009; DÉZIEL et al, 2010; EPASINGHE et al, 2013). Based on the ability to inhibit MMPs, the possibility of this effect for the dentin tissue, aiming the maintenance of the collagen matrix, since this is of paramount importance to minimize the erosion process.

Proanthocyanidin (PA), which has a flavan 3 - ol subunit attached to C4 - C6 (C8), is a natural compound derived from fruits, vegetables, nuts, among others (HAN et al, 2003; KENNEDY, TAYLOR, 2003). Studies have shown that the PA has a high affinity with proteins rich in proline, such as collagen, besides being responsible for increasing the ability of cells to synthesize collagen (HAN et al, 2003).

Other studies have demonstrated a significant improvement in adhesion and mechanical properties at the resin-dentin attachment interface, making adhesive restorations resistant and durable (CASTELLAN et al, 2010; CASTELLAN et al, 2013).

Proanthocyanidin also has showed its effectiveness in preventing dentin erosion because it was able to reduce the dentin wear and DOM degradation when used as gel. (BOTEON et al, 2017) Other studies show that proanthocyanidin-based products, such as gels and dentifrices, may be useful in minimizing the wear of dentin exposed to erosive agents (BUENO et al, 2016; HONÓRIO et al. 2016). Therefore, proanthocyanidin-based home use

products could be extremely useful in preventing large dentin losses by patients who regularly consume acidic beverages. Other extracts have been tested as mouthrinse and have shown good results in reducing dentin wear (MAGALHÃES et al, 2009b). However, mouthrinse based on this specific agent (proanthocyanidin) have never been tested for this purpose.

On the other hand, it is known that fruits rich in proanthocyanidin such as grape and Cranberry have low pH values (Food and Drug Administration FDA – website) and although they have agents that can minimize the degradation of dentin collagen exposed to erosive agents, their low pH values may in turn accelerate this process. Thus, the literature does not clarify whether extracts of proanthocyanidin at its naturally acidic pH will inhibit or induce degradation of the demineralized organic matrix.

Therefore, the aim of this study was to evaluate the efficacy of a proanthocyanidin based naturally acidic mouthrinse and a pH neutral proanthocyanidin based mouthrinse on dentin submitted to erosive challeng

2 DISCUSSION

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The *in situ* study was chosen because it allows reproducing similar conditions of the human oral cavity, contributing to the control of the biases inherent to this type of study and the highly controlled laboratory situation (ZERO, 1995; HONÓRIO et al, 2008). On the other hand, it requires that the work be conducted with great discretion, as it depends on the collaboration of the volunteers (ZERO, 1995; HONÓRIO et al, 2008).

Bovine dentin was the substrate selected for this study because it has been used in several studies (RIOS et al, 2006; HONÓRIO et al, 2008; KATO et al, 2009; HONÓRIO et al, 2010). It presents similarities with human teeth as to its morphology (SCHILKE et al, 2000; FONSECA et al, 2008), physiology (SCHMALZ et al, 2001) and mechanical structure (HARA et al, 2003; WEGEHAUPT et al, 2008). Bovine dentin has already been used in research on dental erosion (KATO et al, 2009; KATO et al, 2010; KATO et al, 2011), and state that this substrate is very reliable for studies involving dentin erosion and the activity of metalloproteinases.

In this study, the agent selected to minimize the progression of dentin erosion was a purified proanthocyanidin from grape seeds. This reduction probably occurs due to inhibition of the MMPs from DOM. The effectiveness of agents has been proven through several studies (HAN et al, 2003; LA, HOWELL, GRENIER, 2009; MAGALHÃES et al, 2009; KATO et al, 2010; CASTELLAN et al, 2011; BEDRAN-RUSSO et al, 2014; KATO et al, 2014; HONÓRIO et al, 2016), with the inhibition effect on DOM collagenolytic enzymes, mainly for chlorhexidine (GENDRON et al, 1999; MAGALHÃES et al, 2009; KIM et al, 2011).

Among the different ways of incorporating active principles, this *in situ* study selected the method of application in the form of a mouthrinse because it allows greater comfort to the patient, since he can use the solution daily without the need to attend the dental office. The use of agents in toothpaste to prevent or control tooth erosion is questionable because brushing may increase damage to softened dental structure after an acid challenge (ATTIN et al, 2001; LUSSI et al, 2014).

In addition, the literature demonstrates the efficacy of rinses when used with different agents in the prevention of initial erosion of the enamel (de OLIVEIRA et al, 2015), in reducing erosion and dentin abrasion (MAGALHÃES et al, 2009b), in the control of dentin

hypersensitivity (MOLINA et al, 2016) and in the protection of deciduous teeth submitted to simulated erosion (M'BAREK et al, 2014).

Fluoride was not part of this study as it has the ability to promote the formation of a metal-rich resistant acid layer or a layer of calcium fluoride that provides temporary protection against erosive challenges. Therefore, to be efficient, this agent would require multiple applications (HUYSMANS, YOUNG, GANSS, 2014).

Chlorhexidine was chosen as a positive control of this work because it showed efficacy in the inhibition of collagen degradation by MMPs and ability to decrease dentin wear (GENDRON et al, 1999; HEBLING et al, 2005; CARRILHO et al, 2007; MAGALHÃES et al, 2009b; KATO et al, 2010; BUZALAF et al, 2012; KATO et al, 2012; BUZALAF, MAGALHÃES, WIEGAND, 2014; ZARELLA et al 2015). An *in situ study* (MAGALHÃES et al, 2009b) in which volunteers performed mouthrinse with chlorhexidine after erosive challenge *ex vivo* showed that it was able to reduce dentin erosion. The literature shows that there are promising results in the inhibition of MMPs when chlorhexidine and green tea are used in the form of a mouthrinse resulting in the reduction of dental erosion (MAGALHÃES et al, 2009b). In addition, the green tea extract used in gel form also showed ability to prevent erosion in dentin (KATO et al, 2010). In this study, the group treated with chlorhexidine and treated with neutralized proanthocyanidin had the best results in the prevention of erosion, with no statistical difference between them. Thus, the use of a neutral proanthocyanidin-based mouthwash at first appears to be more advantageous than chlorhexidine rinse, since it does not present all the side effects presented by the regular use of chlorhexidine, such as: staining of teeth, loss of taste and desquamation of oral mucosae (GUIMARÃES et al, 2006).

Based on that, natural agents have been increasingly studied in the development of oral health products (KHADDAM et al, 2014). Collagen crosslinking agents were tested and the highest statistically significant inhibition of total MMP activity was observed using grape seed extract. Although it did not present a statistically significant difference when compared to the synthetic agents, this work demonstrated that the grape seed extract obtained greater inactivation of the MMPs, proving the best action of natural agents when compared to the synthetic ones (SESEOGLULARI-DIRIHAN et al, 2016).

Among natural agents there is the PA, widely available in fruits, vegetables, nuts, seeds, flowers and barks (FINE, 2000; JOSHI, KUSZYNSKI, BAGCHI, 2001). It is

biocompatible, which presents good bioactivity with dentin and is available as a renewable resource (BEDRAN-RUSSO et al, 2007; PAVAN et al, 2011; BEDRAN-RUSSO et al, 2014).

The literature shows that some agents rich in proanthocyanidin (JOSHI, KUSZYNSKI, BAGCHI, 2001; BEDRAN-RUSSO et al, 2007; BEDRAN-RUSSO et al, 2014) may be able to inhibit the action of MMPs on DOM degradation. The maintenance of DOM is important because the presence of collagen makes it difficult to diffuse the acids into the tissue, thus softening the erosion process (KLETER, 1994).

In vitro [(BOTEON et al, 2017) and *in situ* (HONÓRIO et al, 2014), studies in which the pH of the PA was adjusted, showed that this agent has the ability to prevent the progression of dentin wear. This study did not present the same results by using the extract *in natura* (acid pH), but showed that if pH was adjusted the results are promising. Therefore, the results of the present study have shown that the use of proanthocyanidin-based acid extracts has a more deleterious than beneficial effect.

Although there was decrease in dentin wear, further studies should be conducted to verify whether PA has the potential to bring other benefits when used as mouthrinse and if this concentration is the best one.

3 CONCLUSION

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The results of this study suggest that 10% neutralized proanthocyanidin mouthrinse could be a good strategy to diminish dentin wear progression.

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